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| APPLICATION NO. | FILING DATE | FIRST NAMED INVENTOR | ATTORNEY DOCKET NO. | CONFIRMATION NO. | |
|----------------------|---|----------------------|---------------------|------------------|--|
| 10/568,066 | 02/13/2006 | Wataru Iijima | 286003US0PCT | 7464 | |
| | 7590 08/21/200 AK, MCCLELLAND I | EXAMINER | | | |
| 1940 DUKE ST | REET | HINES, LATOSHA D | | | |
| ALEXANDRIA, VA 22314 | | | ART UNIT | PAPER NUMBER | |
| | | 4112 | | | |
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Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

Notice of the Office communication was sent electronically on above-indicated "Notification Date" to the following e-mail address(es):

patentdocket@oblon.com oblonpat@oblon.com jgardner@oblon.com

| | Application No. | Applicant(s) | | | | |
|--|--|---|--|--|--|--|
| | 10/568,066 | IIJIMA ET AL. | | | | |
| Office Action Summary | Examiner | Art Unit | | | | |
| | LATOSHA HINES | 4112 | | | | |
| The MAILING DATE of this communication ap Period for Reply | opears on the cover sheet with the o | correspondence address | | | | |
| A SHORTENED STATUTORY PERIOD FOR REPLY WHICHEVER IS LONGER, FROM THE MAILING IDENTIFY OF THE MONTHS FROM THE MAILING IDENTIFY OF THE MONTHS FROM THE MAILING IDENTIFY OF THE MONTH OF THE M | DATE OF THIS COMMUNICATION .136(a). In no event, however, may a reply be tilt d will apply and will expire SIX (6) MONTHS from the, cause the application to become ABANDONE | N. mely filed the mailing date of this communication. ED (35 U.S.C. § 133). | | | | |
| Status | | | | | | |
| Responsive to communication(s) filed on 13 and 2a) This action is FINAL . 2b) The 3) Since this application is in condition for allowed closed in accordance with the practice under | is action is non-final. ance except for formal matters, pro | | | | | |
| Disposition of Claims | | | | | | |
| 4) Claim(s) 1-6 is/are pending in the application 4a) Of the above claim(s) is/are withdres 5) Claim(s) is/are allowed. 6) Claim(s) 1-6 is/are rejected. 7) Claim(s) is/are objected to. 8) Claim(s) are subject to restriction and/ Application Papers 9) The specification is objected to by the Examination of the drawing(s) filed on is/are: a) and according to a position of the drawing(s) filed on is/are: a) and according to a position of the above claim(s) are subjected to by the Examination of the drawing(s) filed on is/are: a) according to a position of the application | awn from consideration. or election requirement. | Examiner. | | | | |
| Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a). Replacement drawing sheet(s) including the correction is required if the drawing(s) is objected to. See 37 CFR 1.121(d). 11) The oath or declaration is objected to by the Examiner. Note the attached Office Action or form PTO-152. | | | | | | |
| Priority under 35 U.S.C. § 119 | | | | | | |
| 12) Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f). a) All b) Some * c) None of: 1. Certified copies of the priority documents have been received. 2. Certified copies of the priority documents have been received in Application No 3. Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)). * See the attached detailed Office action for a list of the certified copies not received. | | | | | | |
| Attachment(s) 1) Notice of References Cited (PTO-892) 2) Notice of Draftsperson's Patent Drawing Review (PTO-948) 3) Information Disclosure Statement(s) (PTO/SB/08) Paper No(s)/Mail Date 02/13/2006. | 4) Interview Summary Paper No(s)/Mail D 5) Notice of Informal F 6) Other: | ate | | | | |

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DETAILED ACTION

1. This is the initial Office action based on the 10/568066 application filed on February 13, 2006.

2. Claims 1-6 are pending and have been fully considered.

Claim Rejections - 35 USC § 102

3. The following is a quotation of the appropriate paragraphs of 35 U.S.C. 102 that form the basis for the rejections under this section made in this Office action:

A person shall be entitled to a patent unless -

(e) the invention was described in a patent granted on an application for patent by another filed in the United States before the invention thereof by the applicant for patent, or on an international application by another who has fulfilled the requirements of paragraphs (1), (2), and (4) of section 371(c) of this title before the invention thereof by the applicant for patent.

The changes made to 35 U.S.C. 102(e) by the American Inventors Protection Act of 1999 (AIPA) and the Intellectual Property and High Technology Technical Amendments Act of 2002 do not apply when the reference is a U.S. patent resulting directly or indirectly from an international application filed before November 29, 2000. Therefore, the prior art date of the reference is determined under 35 U.S.C. 102(e) prior to the amendment by the AIPA (pre-AIPA 35 U.S.C. 102(e)).

4. Claim 6 is rejected under 35 U.S.C. 102(e) as being clearly anticipated by **SAKA** (US 2006/0025620).

With respect to claim 6 **SAKA** discloses a the method for producing a fatty acid alkyl ester composition capable of being utilized effectively as a diesel fuel (particularly bio-diesel fuel) by treating fats and oils containing a fatty acid glyceride and/or fatty acid (paragraph 0018-0019). The "fats and oils" means those containing a fatty acid glyceride and/or fatty acid as described above

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contain a main component called fatty acid mono-glyceride, fatty acid diglyceride, fatty acid tri-glyceride. Page 3

Claim Rejections - 35 USC § 103

- 5. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:
 - (a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.
- 6. Claims 1-4 are rejected under 35 U.S.C. 103(a) as being unpatentable over **SAKA (US 2006/0025620)**.

With respect to claims 1 and 2 **SAKA** discloses the method for producing a fatty acid alkyl ester composition capable of being utilized effectively as a diesel fuel (particularly bio-diesel fuel) by treating fats and oils containing a fatty acid glyceride and/or fatty acid (paragraph 0018-0019). Alcohol (preferably methanol) and/or water, which works as an acid catalyst (paragraph 0022) are allowed to co-exist with the above mentioned fats and oils and the reaction is conducted under conditions of a temperature of 100 °C to 370 °C and a pressure of 1 to 100 MPa for up to 20 minutes (paragraph 0017 and 0021). **SAKA** discusses the removal of the unreacted methanol and water and by-product of glycerin from the hydrolysis of tri-glyceride and esterfication of a fatty acid. **SAKA** conducted various tests that included three different embodiments with five different examples. In Example 5, the esterification and transesterification of a fatty acid

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is conducted. For the transesterification a rapeseed oil is used at a reaction temperature of 300 °C and 350 °C for 15 minutes. Table 8 shows the results:

TABLE 8

| | Critical temperature | Critical pressure | Pressure in example (MPa) | |
|------------|-------------------------|----------------------|------------------------------|---------|
| Alcohol | (° C.) | (Mpa) | 300° C. | 350° C. |
| Methanol | 239 | 8.09 | 20 | 43 |
| Ethanol | 243 | 6.38 | 15 | 25 |
| l-proparol | 264 | 5.06 | 10 | 23 |
| i-bumanoi | 287 | 4.90 | 9 | 23 |
| l-octanol | 385 | 2.86 | 5 | 29 |

with the temperature at 350 °C and the pressure at 43 MPa for the reaction of a rapeseed oil and methanol It is obvious to one of ordinary skill the results will vary when used at higher temperatures and pressures for a shorter amount of time which causes the catalyst to degrade at a faster rate. The discovery of an optimum value of a known result effective variable, without producing any new or unexpected results, is within the ambit of a person of ordinary skill in the art. See *In re Boesch*, 205 USPQ 215 (CCPA 1980) (See MPEP 2144).

With respect to claim 3 **SAKA** discloses an example of esterification of a fatty acid group and transesterification of rapeseed oils occurring parallel to each reaction. Table 7 shows the result:

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TABLE 7

| Example | Aicobol/fats and oils (mole ratio) | Fats and ods | Aicobol | Temperature (° C.) | Pressure (Mpa) | Resction time (mis) | Yleid (%) |
|-----------------|--|-----------------------|------------|-----------------------|-------------------|------------------------|--------------|
| Example 5-1 | 42/1 | C ₂₉₋₃ | Methanol | 300 | 20 | 8 | 96.2 |
| Example 5-2 | 42/1 | C_{28-2} | | 300 | 20 | 8 | 95.1 |
| Example 5-3 | 42/1 | $C_{s,\theta+\delta}$ | | 300 | 20 | 8 | 95.8 |
| Example 5-4 | 42/1 | C:s-0 | | 300 | 20 | 8 | 94.7 |
| Example 5-5 | 42/1 | C ₂₅₋₀ | | 300 | 20 | 8 | 94,0 |
| Example 5-5 | 42/1 | Rispesced oil | | 300 | 20 | 15 | 98.0 |
| Example 5-7 | 42/1 | Rapesced oil | | 350 | E # | 4 | 9 ₹.8 |
| Example 5-8 | 42/1 | © _{25–3} | Ethanoi | 300 | 15 | 12 | 94.6 |
| Example 5-9 | 42/1 | C ₁₈₋₂ | | 300 | 15 | 14 | 97.4 |
| Example 5-10 | 4 2/1 | C _{3,8-3} | | 300 | 15 | 14 | 95.9 |
| Example 5-11 | 42/1 | C⁵a−2 | | 300 | 15 | 15 | 91,2 |
| Example 5-12 | 42/1 | C:s-c | | 300 | 15 | 14 | 91.7 |
| Example 5-13 | 42/1 | Rapeseed oil | | 300 | 15 | 45 | 96.7 |
| Example 5-14 | 42/1 | Rapeseeti oil | | 350 | 25 | 10 | 97.1 |
| Example 5-15 | 4 2/1 | C _{5.8-3} | l-propanol | 300 | 10 | 15 | 9 7.0 |
| Example 5-16 | 4 2/1 | C:3-3 | | 300 | 10 | 14 | 92.7 |
| Example 5-17 | 42/1 | C _{2:5-2} | | 300 | 10 | 14 | 92.3 |
| Example 5-18 | 42/1 | C ₂₉₋₀ | | 300 | 10 | 14 | 89.6 |
| Example 5-19 | 42/1 | C _{is-o} | | 300 | 10 | 14 | 90.1 |
| Example 5-20 | 42/1 | Repeseed oil | | 300 | 10 | 45 | 95.1 |
| Example 5-21 | 42/1 | Rapeseed cil | | 350 | 23 | 24 | 98.8 |
| Example 5-22 | 42/1 | C ₁₈₋₃ | i-butonai | 300 | 9 | 15 | 97.3 |

TABLE 7-continued

| Example | Alcohol/fets and oils (mole ratio) | Fars and oils | Aiccircl | Temperature (° C.) | Pressure (Mp2) | Resotion time (min) | Yleid (%) |
|-----------------|--|-------------------|-----------|-----------------------|-------------------|------------------------|-----------|
| Example 5-23 | 42/1 | C 33-2 | | 3 00 | ŷ | 14 | 92.4 |
| Example 5-24 | 42/1 | C ₃₈₋₃ | | 300 | 9 | 14 | 85.1 |
| Example 5-25 | 42/1 | C ₁₈₋₀ | | 300 | 3 |] दं | 82.5 |
| Example 5-26 | 42 /1 | C _{id=0} | | 300 | 9 | 14 | 81.1 |
| Example 5-27 | 42/1 | Rapeseed oil | | 300 | 9 | 45 | 57.1 |
| Example 5-28 | 42/3 | Rapessed oil | | 350 | 23 | 14 | 95.3 |
| Example 5-29 | 42/1 | Repessed oil | 1-octanel | 300 | 6 | 45 | 68.7 |
| Example 5-30 | 42/1 | Repeseed oii | | 350 | 19 | 2 0 | 96.7 |

 C_{16-6} : palmitic acid,

C₁₈₋₀: steam asid,

C₁₃₋₁; ciese soid, C₁₃₋₂; linoicic soid,

C₁₈₋₃: linclenic acid

It is obvious to one of ordinary skill the results will vary when used at higher temperatures and pressures for a shorter amount of time which causes the catalyst to degrade at a faster rate. The discovery of an optimum value of a known result effective variable, without producing any new or unexpected results, is within the ambit of a person of ordinary skill in the art. See In re Boesch, 205 USPQ 215 (CCPA 1980) (See MPEP 2144).

With respect to claim 4 **SAKA** discloses a fatty acid and methanol under conditions of volume ratio, temperature, and pressure. Table 5 shows results: Art Unit: 4112

TABLE 5

| Example | Fatty scid | Famy acid (mi)/ methanol (mi) | Tem- perature (° C.) | Pressure (Mps) | Resc- tion time (min) | Yield (%) |
|--------------|-----------------------|----------------------------------|----------------------------|-------------------|--------------------------------|--------------|
| Tryembia | 2620 | menumor (mm) | (V .,} | έπετħα\ | (1111111) | £203 |
| Example 3-1 | $C_{2,6-3}$ | 0.91:4.09 | 270 | 17 | 20 | 90 |
| Example 3-2 | ರ್ಚಿಕ್ತ | 0.91:4.09 | 300 | 24 | 7 | 88 |
| Example 3-3 | ್ಯೄ_ | 3.91:4.09 | 350 | 43 | 4 | 75 |
| Comparative | ⊏್ಯ⊶ | 3.9154.09 | 4 00 | 75 | 2 | 92 |
| example 3-1 | 2.00-20 | | | | | |
| Example 3-4 | ದ್ಯೄ_ರ | 3.9154.09 | 270 | 17 | 20 | 98 |
| Example 3-5 | ೭ೄ್ತಿ | 0.91:4.09 | 300 | 24 | 7 | 98 |
| Example 3-6 | ℂೄತ್ತಿ | 3.91:4.09 | 350 | 43 | 4 | 100 |
| Comparative | ದ್ವಿತ್ತವ | 3.91:4.09 | 400 | 75 | 2 | 100 |
| example 3-2 | 4.4-2 | | | | | |
| Example 3-7 | $C_{i,\mathbf{s}-i}$ | 0.91:4.09 | 270 | 17 | 20 | 98 |
| Example 3-8 | C ₃₈₋₃ | 0.91:4.09 | 300 | 24 | 7 | 98 |
| Example 3-9 | C ₃₈₋₃ | 3.9154.09 | 350 | 43 | 4 | 98 |
| Comparative | C_{18-3} | 9.91:4.09 | 400 | 75 | 2 | 94 |
| example 3-3 | - 2.9-2 | | | | | |
| Example 3-10 | $C_{8\mathbf{z}=2}$ | 0.91:4.69 | 270 | 17 | 20 | 98 |
| Example 3-11 | C₃ _{8–2} | 0.91:4.09 | 300 | 24 | 7 | 98 |
| Example 3-12 | C ₂₈₋₂ | 0.91:4.09 | 350 | 43 | 4 | 87 |
| | | | | | | |

with the fatty acid (ml)/ methanol (ml) ration of 0.91 to 4.09. It is obvious to one of ordinary skill the results will vary when used at higher temperatures and pressures for a shorter amount of time which causes the catalyst to degrade at a faster rate. The discovery of an optimum value of a known result effective variable, without producing any new or unexpected results, is within the ambit of a person of ordinary skill in the art. See *In re Boesch*, 205 USPQ 215 (CCPA 1980) (See MPEP 2144).

7. Claim 5 is rejected under 35 U.S.C. 103(a) as being unpatentable over **SAKA** (US 2006/0025620) in view of SIE et al. (US4,868,221).

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With respect to claim 5 **SAKA** does not disclose a methanolysis reaction carried out in a Hastelloy reaction tube.

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However, with respect to claim 5 <u>SIE et al.</u> discloses a process for the preparation of methanol which comprises contacting gaseous mixtures comprising of carbon monoxide and hydrogen with a catalytic system prepared by a nickel salt, an alcohol, and a hydride alkali metal (column 1 lines 39-49). The methanol produced may be used for a variety of purposes such as: manufacture of synthetic gasoline, as a fuel component, and for the production of methyl tert-butyl ether (column 4 lines 1-5). The experiment was carried out in a 300 mL Hastelloy C autoclave with sodium hydride used as a suspension in white paraffin oil (column 4 lines 28-35).

At the time of the invention it would have been obvious to one of ordinary skill in the art to add a Hastelloy reaction tube invention of **SAKA**. The motivation to do so is the Hastelloy reaction tube helps in temperature control and pressure as taught by **SIE et al.** (column 4 lines 36-56).

Conclusion

8. The prior art made of record and not relied upon is considered pertinent to applicants disclosure.

The reference <u>HAAS et al. (US 6,399,800)</u> discloses achieving highly efficient fatty acid alkyl ester synthesis using soapstock (from crude vegetable oils) or other mixtures of vegetable lipids (column 3 lines 5-17).

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9. Any inquiry concerning this communication or earlier communications from the

examiner should be directed to LATOSHA HINES whose telephone number is

(571)270-5551. The examiner can normally be reached on Monday thru Thursday and

alternate Fridays from 8 a.m. to 5 p.m.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's

supervisor, Barbara Gilliam can be reached on 571-272-1330. The fax phone number

for the organization where this application or proceeding is assigned is 571-273-8300.

Information regarding the status of an application may be obtained from the

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USPTO Customer Service Representative or access to the automated information

system, call 800-786-9199 (IN USA OR CANADA) or 571-272-1000.

LH

20080815

/Brian Sines/

Primary Examiner, Art Unit 1797